

Photosynthesis



Life in the Sun

- Light is central to the life of a plant
- Photosynthesis is the most important chemical process on Earth
 - It provides food for virtually all organisms
- Plant cells convert light into chemical signals that affect a plant's life cycle



Energy Storage and Transformation

2 Types of processes

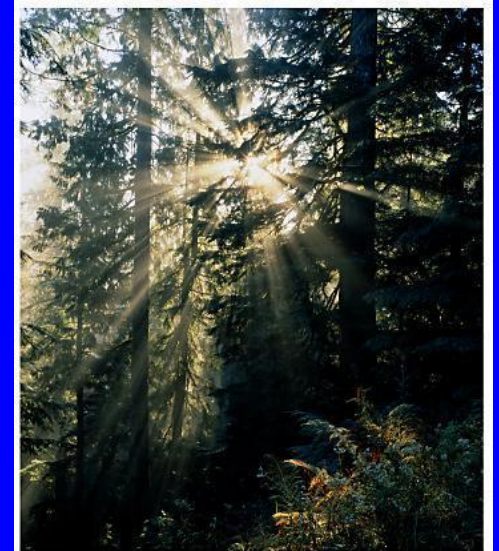
Photosynthesis- solar energy is converted to chemical bond energy within carbohydrates

Cellular respiration- carbohydrates are converted into useable forms of energy such as ATP, which provides energy for various different systems. E.g. synthesis of chemicals, active transport, and contraction of muscle fibers

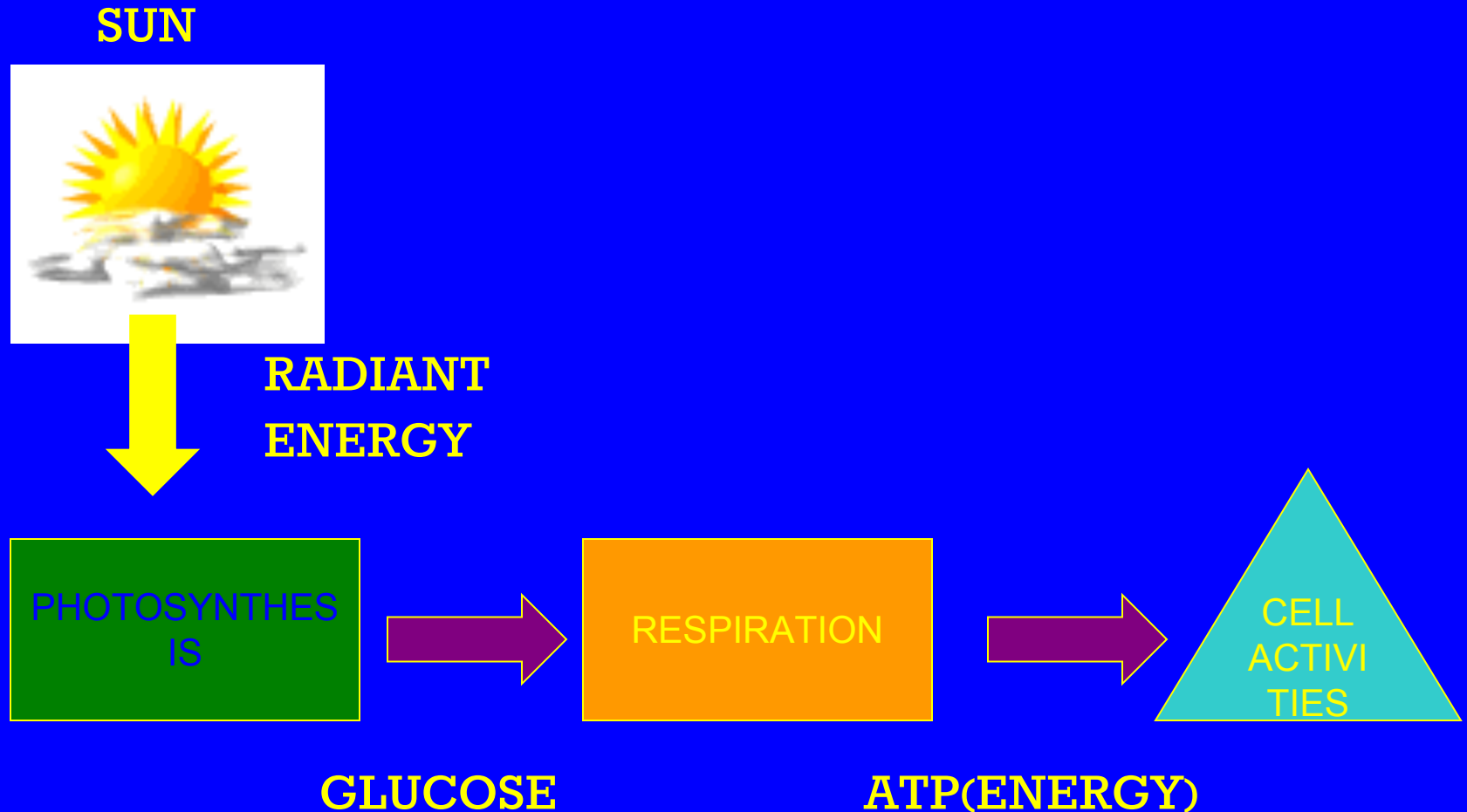
Life on Earth is solar powered

On a global scale the productivity of photosynthesis is astounding

Photosynthesis produces 160 billion metric tons of carbohydrates each year



Overview of photosynthesis and respiration



Terms

1st law of thermodynamics

Energy cannot be created or destroyed, but energy can change forms. The amount of energy within a closed system remains constant.

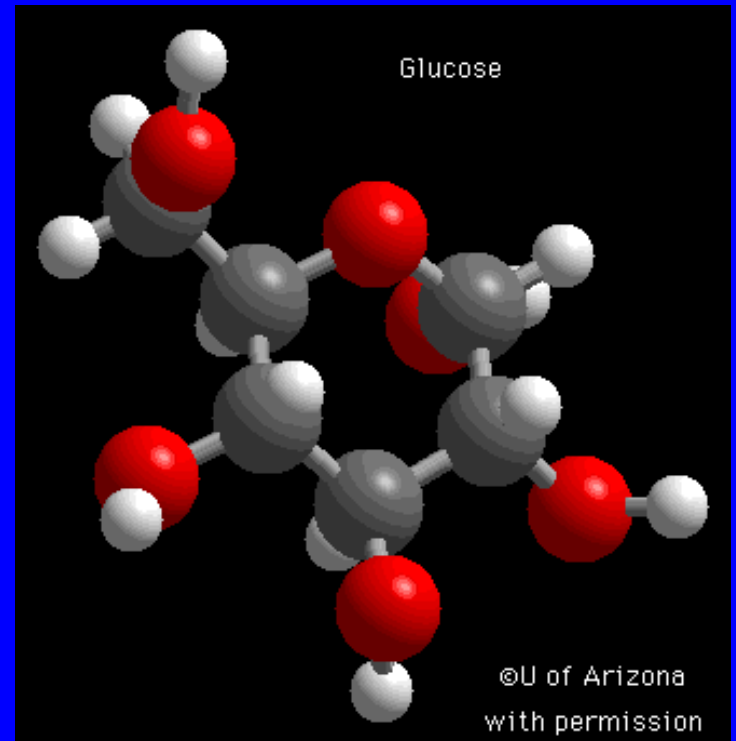
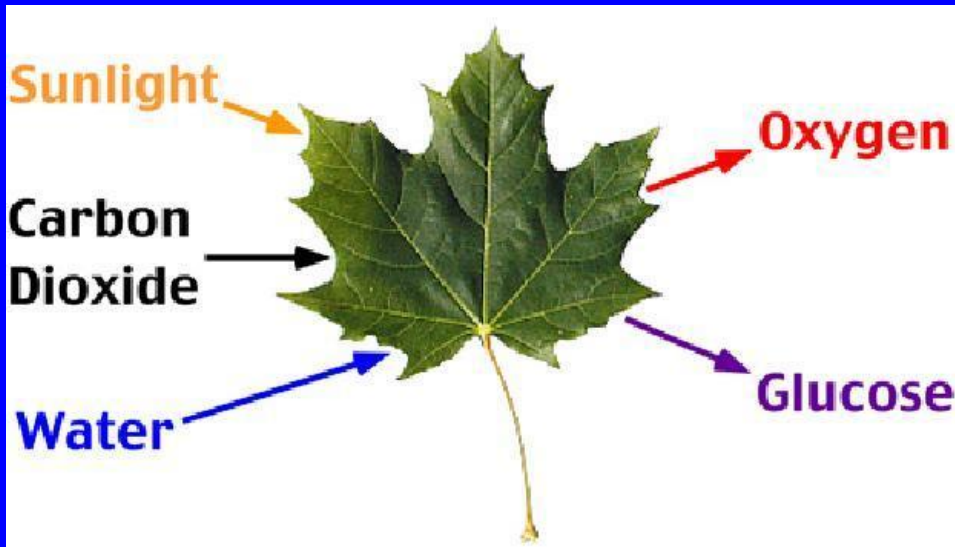
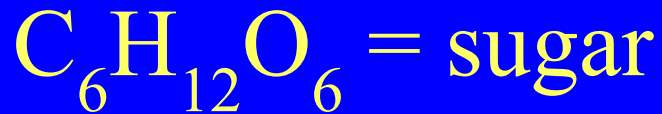
2nd law of thermodynamics

All conversions of energy produce some heat, which is not useful energy. The amount of energy that is unavailable for work is referred to as entropy.

- I. Photosynthesis Stores Energy as Carbohydrate.
- A. Light energy is stored in plants when photosynthesis converts inorganic compounds to organic.
 - B. Glucose is the organic storage form of chemical energy. (Savings!)
 - C. The energy from glucose is converted into ATP in cell respiration. (Cash!)
 - D. Photosynthesis produces glucose.

E. The equation for photosynthesis is:

Light



- Stomata, open and closed
- Site of gas exchange O_2 , H_2O and CO_2

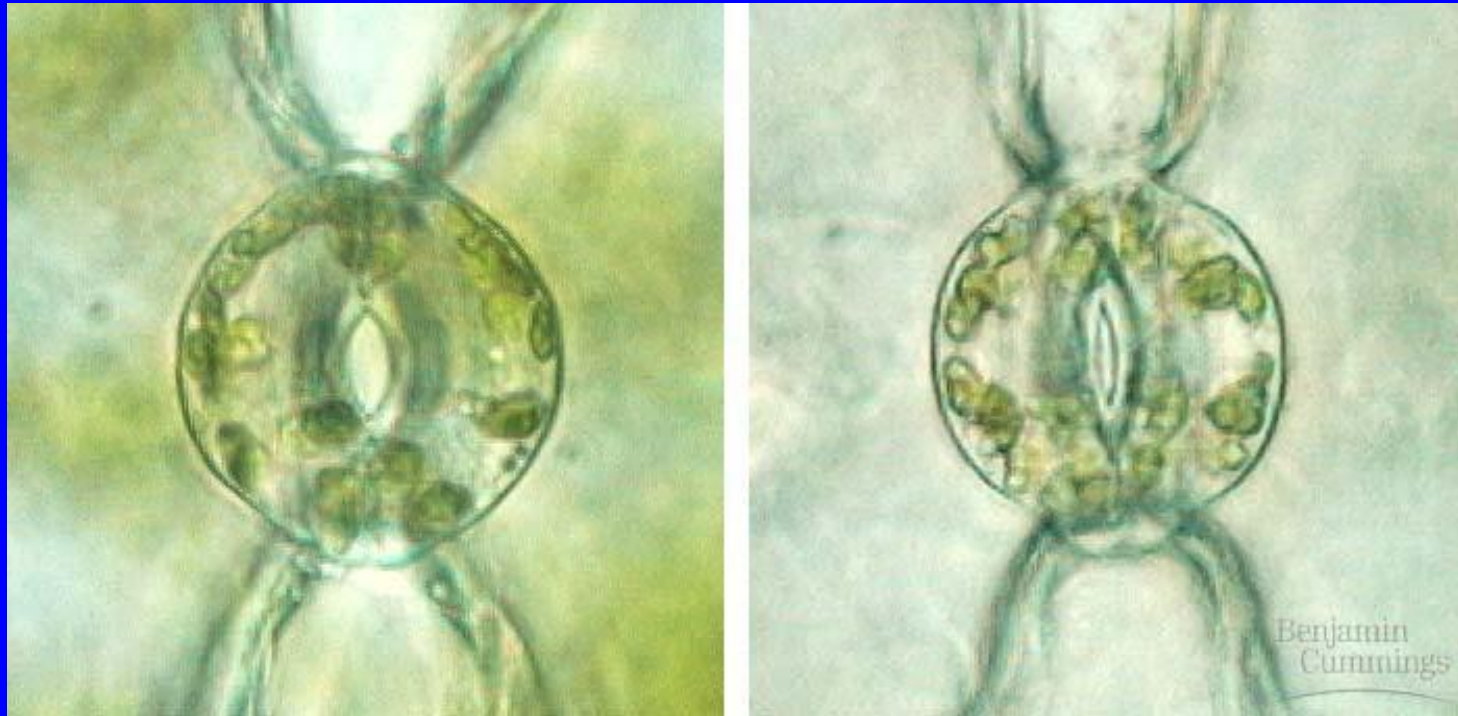


Figure 7.2x2

This reaction occurs inside of plants, and specifically within the leaves. When light reaches the surface of a green leaf one of three things can happen to the light:

- Reflected back off of the leaf
- Transmitted right through the leaf
- Absorbed by the leaf, by chlorophyll which is found in the chloroplasts

- The full range of radiation is called the electromagnetic spectrum

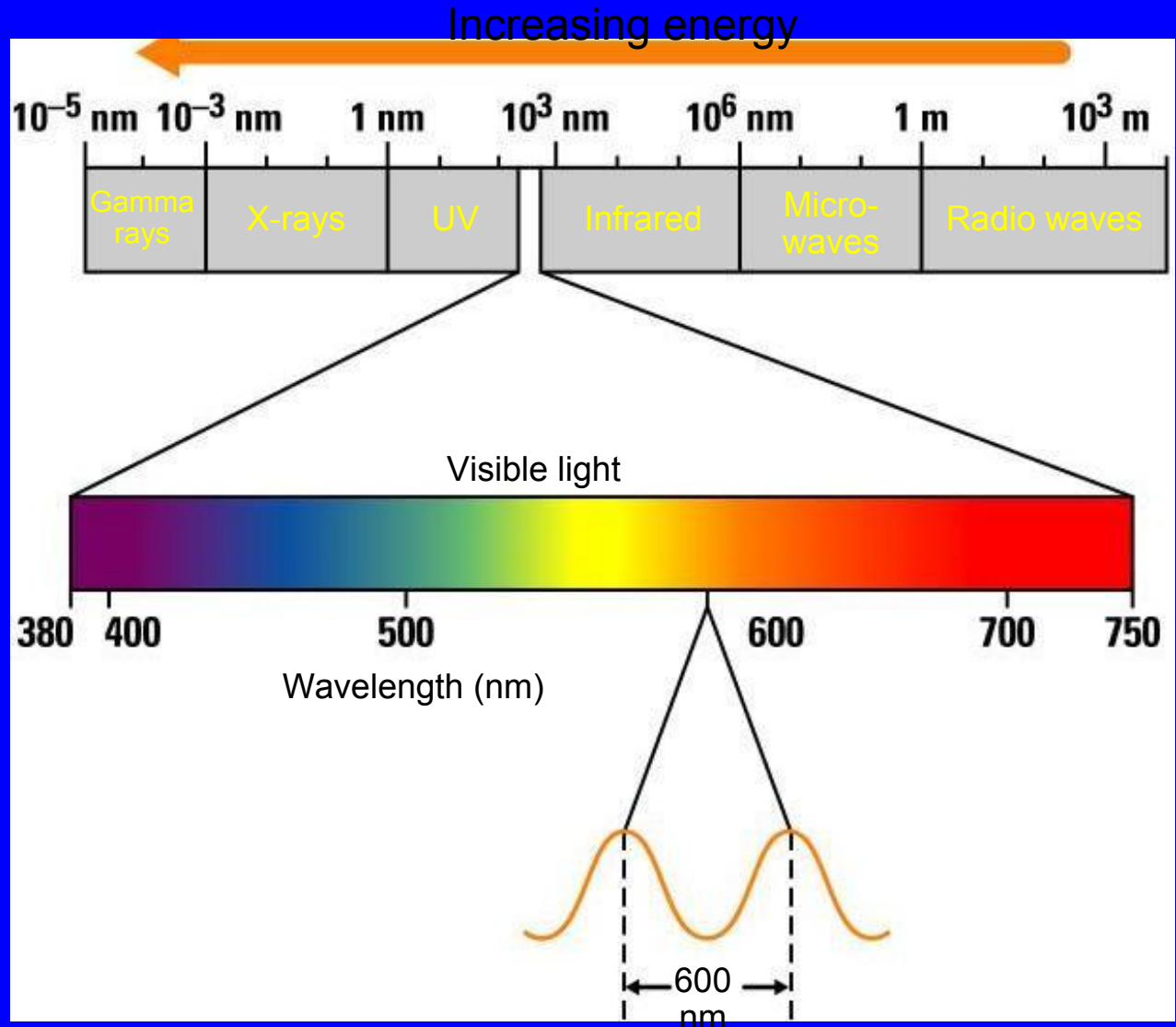


Figure 7.5

The color of a pigment is determined by the wavelengths of light that are reflected. There are at least 6 different types of chlorophyll found in photosynthetic plants with chlorophyll a and b being most common and present in most green land plants. These 2 chlorophylls absorb light chiefly of the violet, blue, orange, and red wavelengths.

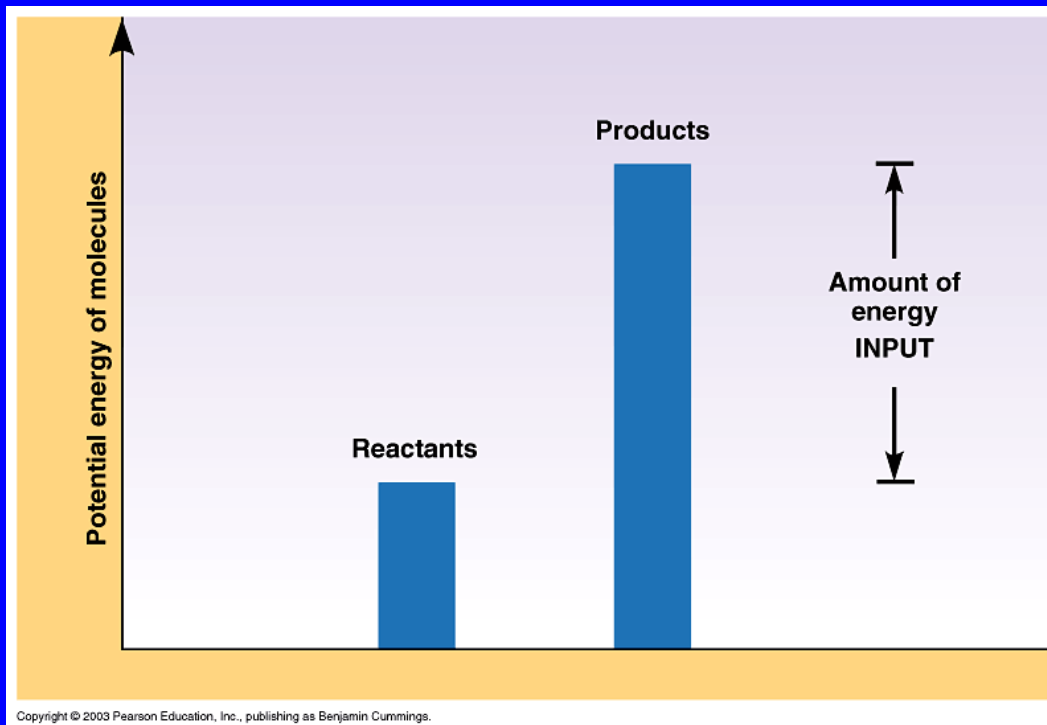
- In the fall, chlorophyll breaks down due to a lack of soil nitrogen. This makes the accessory pigments visible.



Terms to Know

Endergonic Reactions

- Requires energy
- Products have more energy



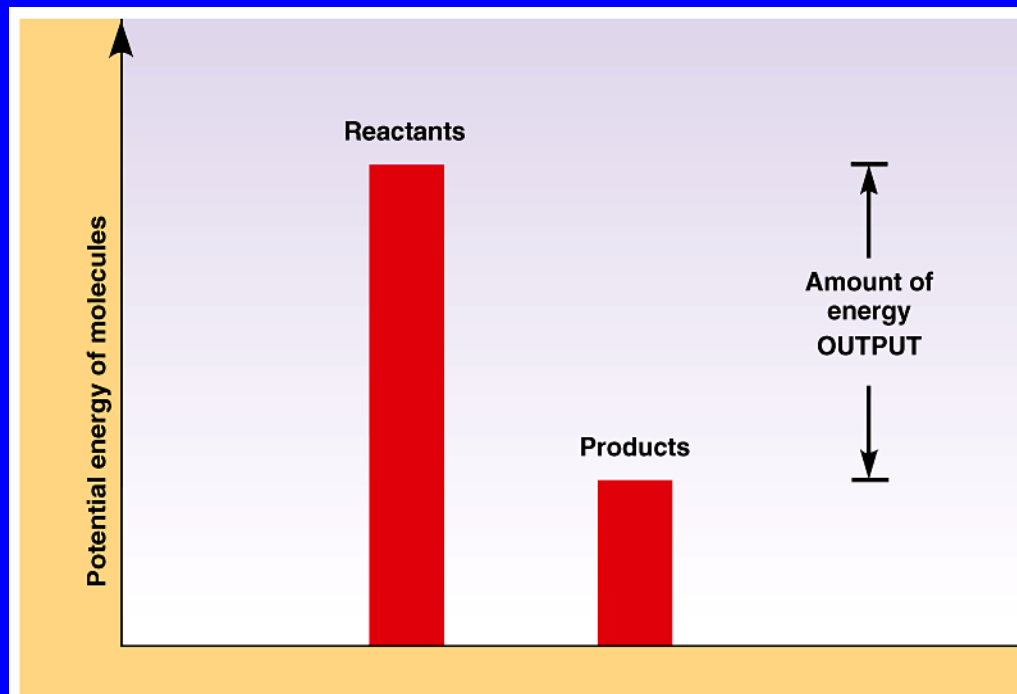
Exergonic Reactions

Releases energy

- Products have less energy
- Ex. Burning

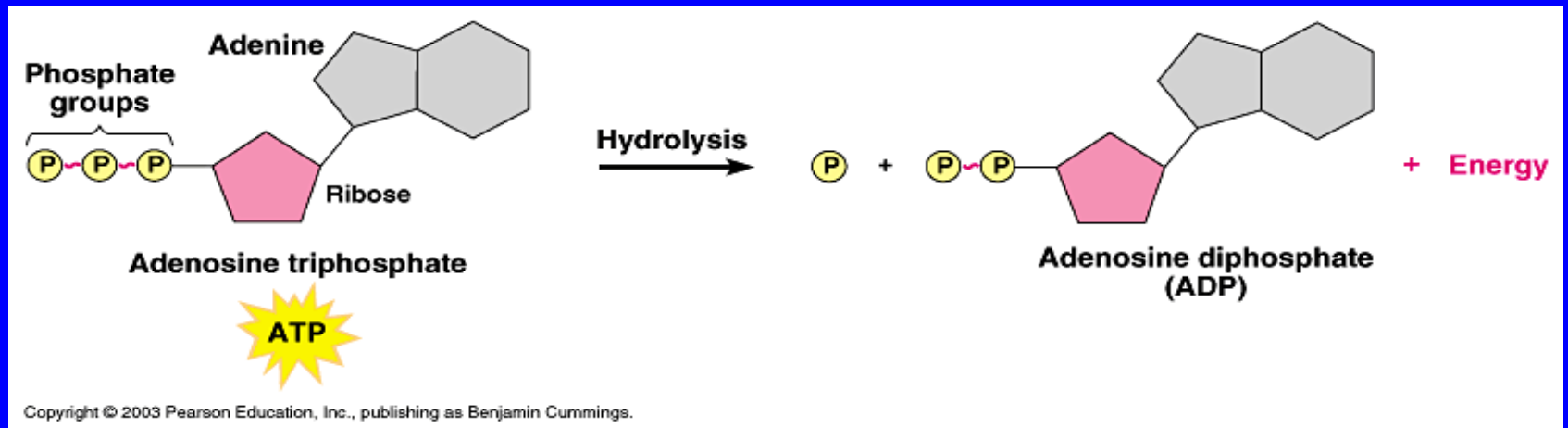
Cellular metabolism

- All reactions in a cell



Terms to know:

ATP (adenosine triphosphate) is a useable form of chemical energy. ATP is a five-carbon sugar (ribose), a nitrogen base (adenine), and 3 phosphate molecules.



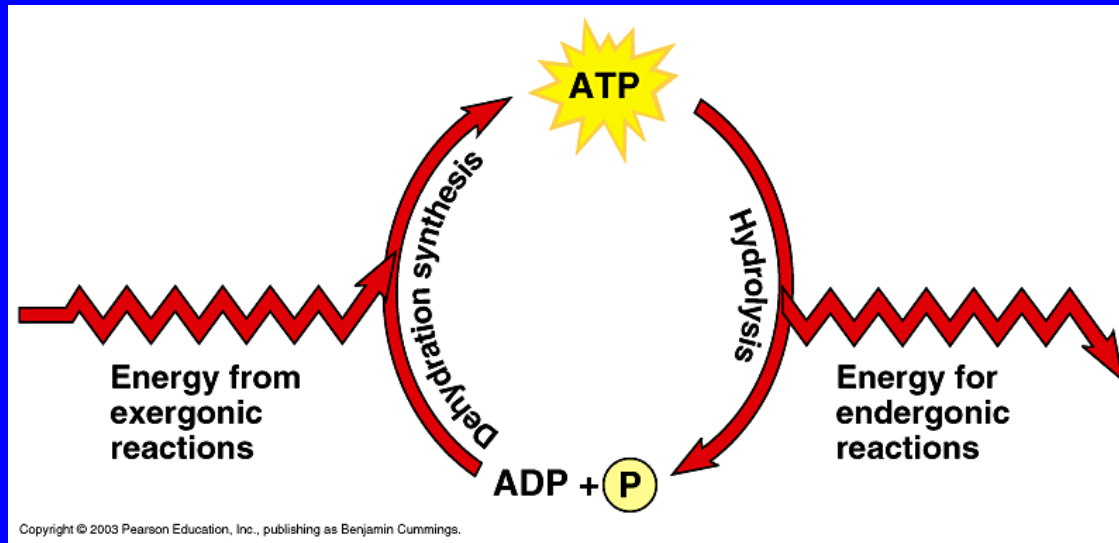
Can be thought of as a gold coin that cells use to pay for work.

When energy is required for cell functions special enzymes extract a phosphate molecule to produce ADP (adenosine diphosphate) and energy.



Which can be reduced even further to AMP (adenosine monophosphate) and energy. However, the bond energy is not as great, so you don't get as good of a bang for your buck.

- ATP cycle
 - Release of third phosphate from ATP
 - Makes energy available to do work
 - ATP becomes ADP (diphosphate)
 - When energy released from food can add phosphate to ADP making ATP



The supply of ATP must be constantly replenished by phosphorylation (the addition of phosphate molecules). This energy production in cells takes place within the electron transport system.

Some elements are strong electron donors and tend to lose their electrons to other elements to become stable. Other elements want to accept these extra electrons to become stable.

The transfer of electrons from one element to another produces more stable ions or compounds and releases energy, which is used to make ATP.

II. Oxidation Reduction Reactions

A. Electrons can carry energy with a new compound.

B. When a molecule loses an

C. When a molecule gains an electron

D. Photosynthesis is where electrons sugars by reduction.

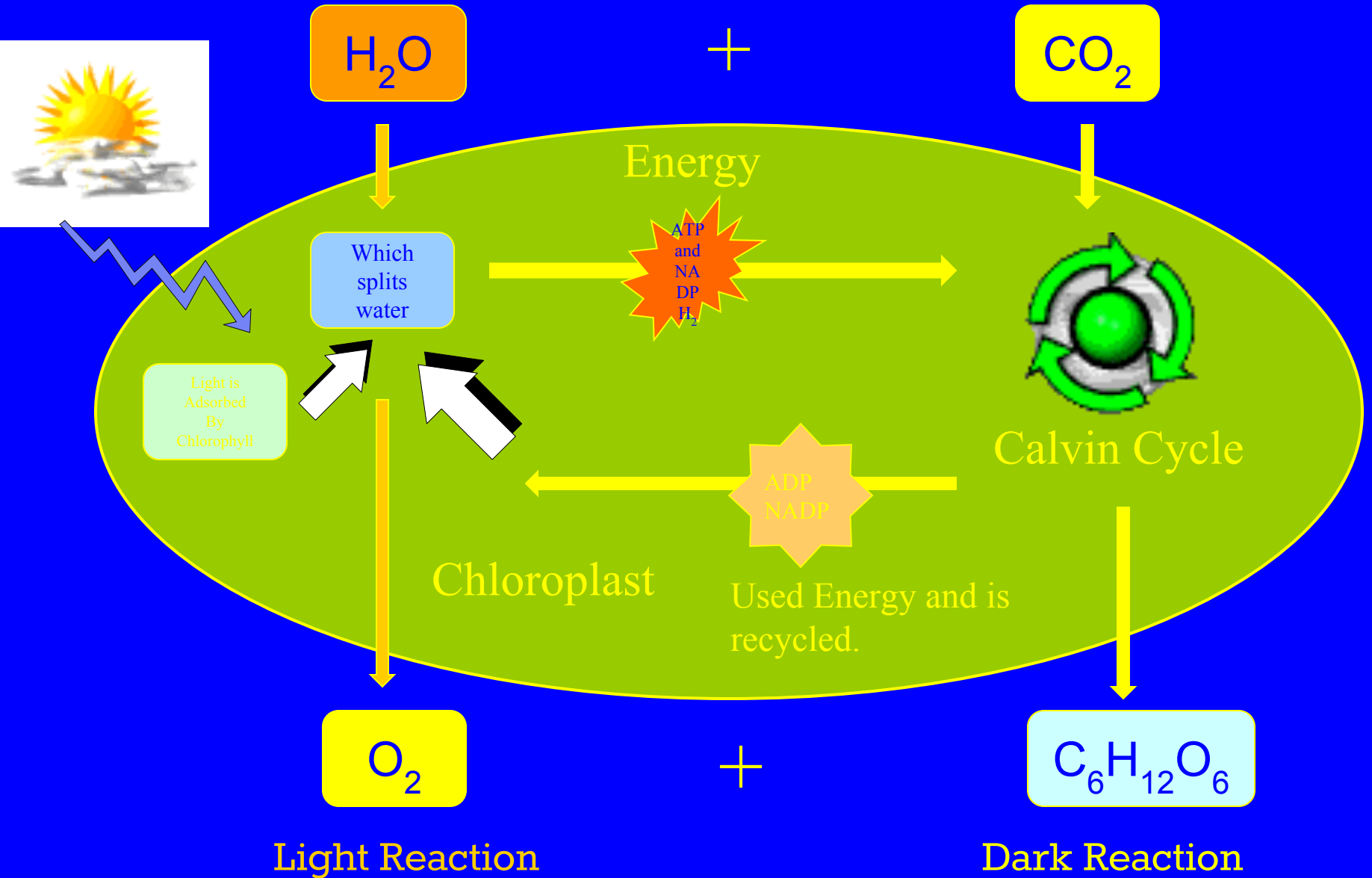
them when they are transferred to

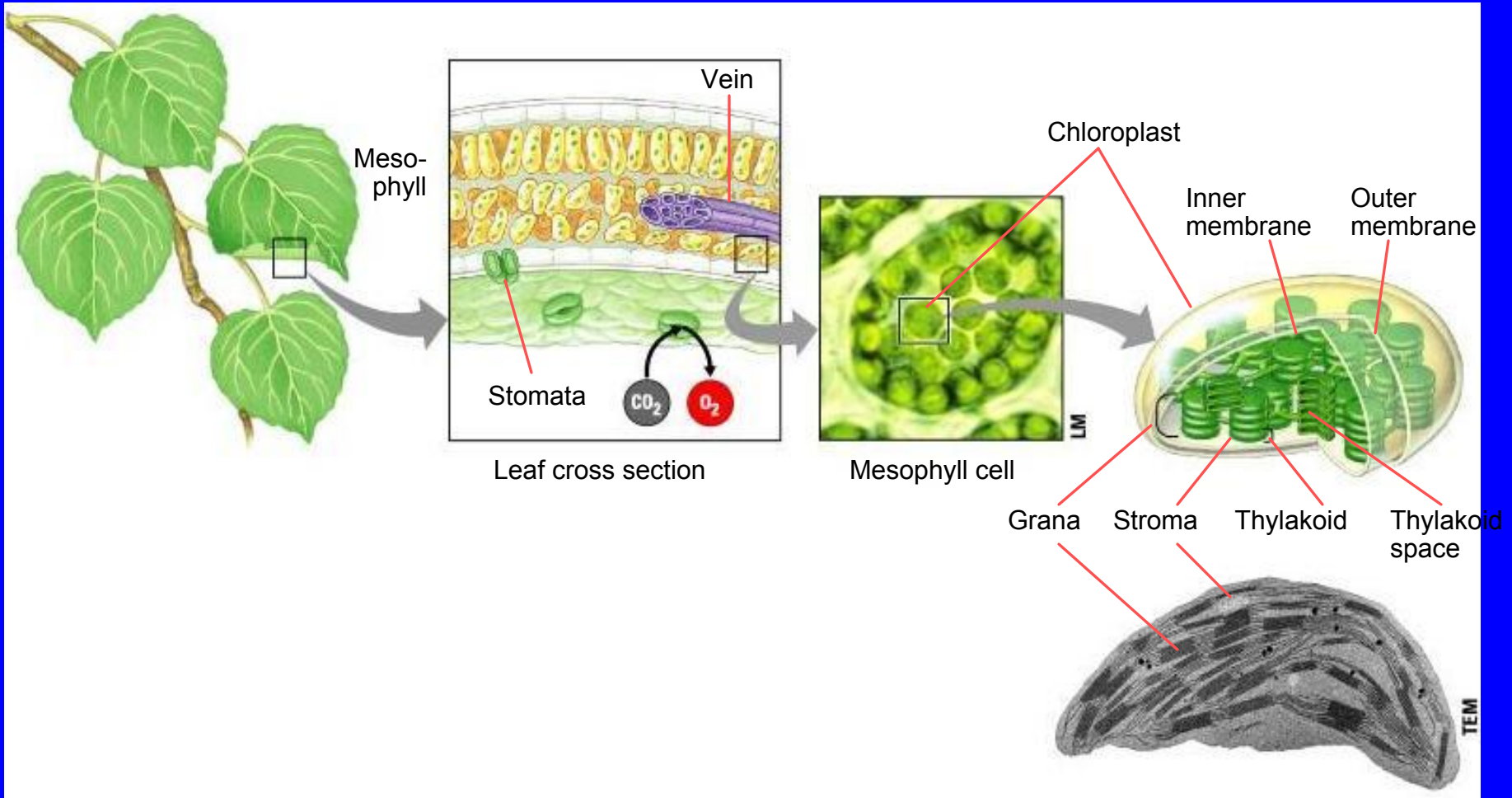
electron it is oxidized. (OIL)

it is reduced. (RIG)

carry energy from sunlight into

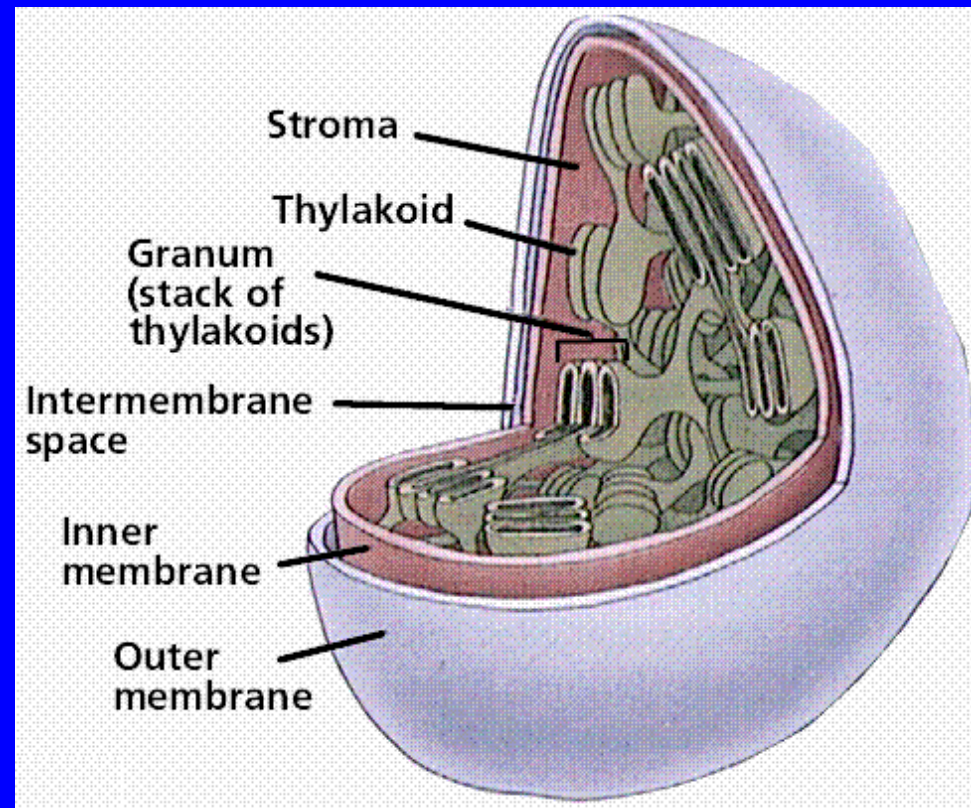
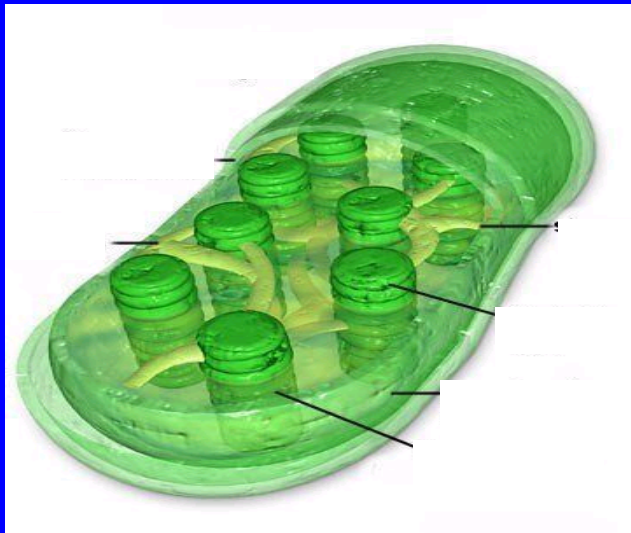
Photosynthesis





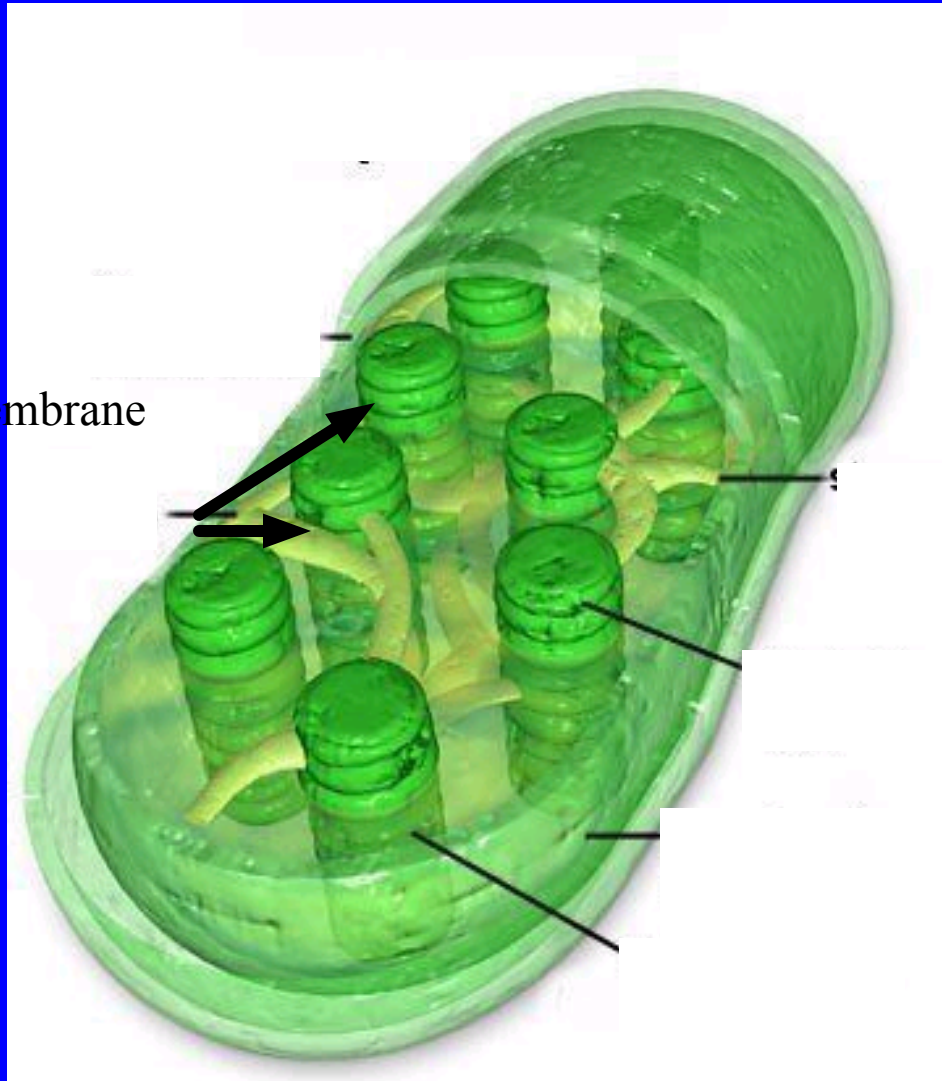
Photosynthesis occurs in the chloroplast.

Chloroplasts contain many granum, which are composed of many thylakoids stacked on top of each other like pancakes. The space surrounding the granum is known as the stroma. Light Reactions occur in the grana and Dark Reactions take place in the stroma of chloroplasts. Light is captured by chlorophyll, which is the pigment that makes plants green.



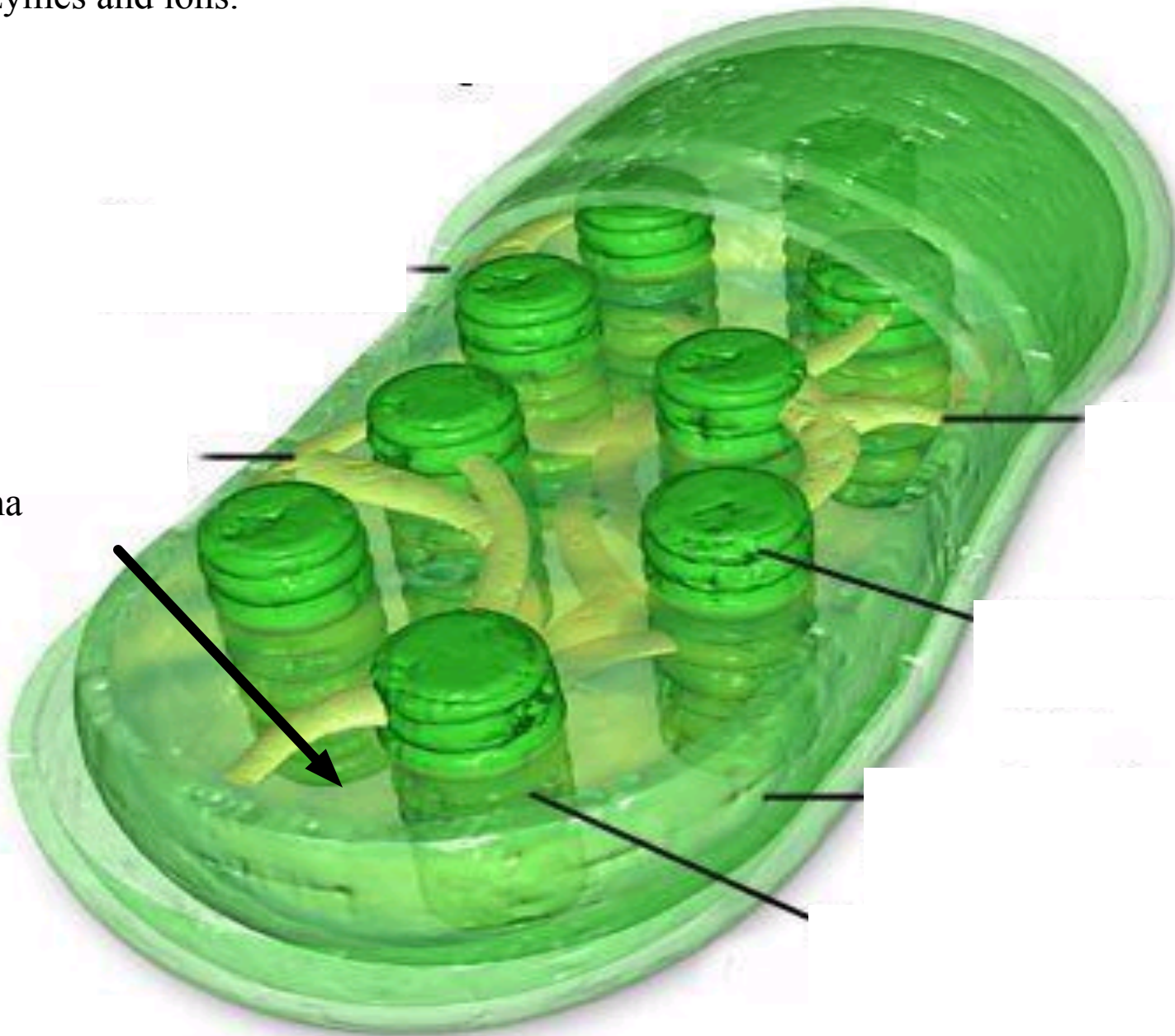
a. The chloroplast has a double outer membrane.

Double Membrane

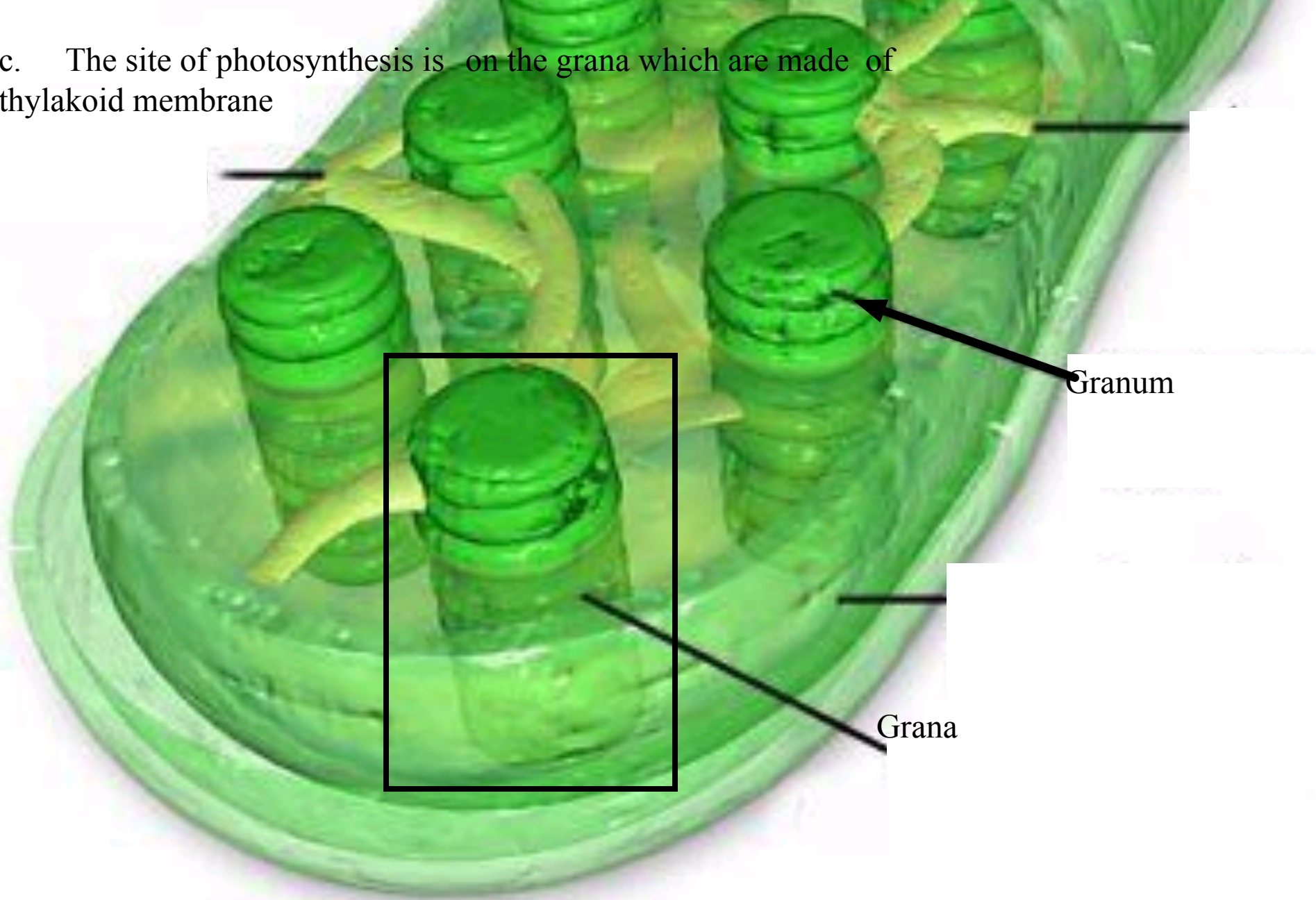


- b. The chloroplast is filled with stroma which is a fluid that contains enzymes and ions.

Stroma



c. The site of photosynthesis is on the grana which are made of thylakoid membrane



In simplified terms, photosynthesis occurs when CO_2 (carbon dioxide) is absorbed into the air spaces of a leaf primarily through diffusion. This CO_2 combines with H_2O (water) in the presence of light energy to produce a sugar molecule (glucose), water, and free oxygen.

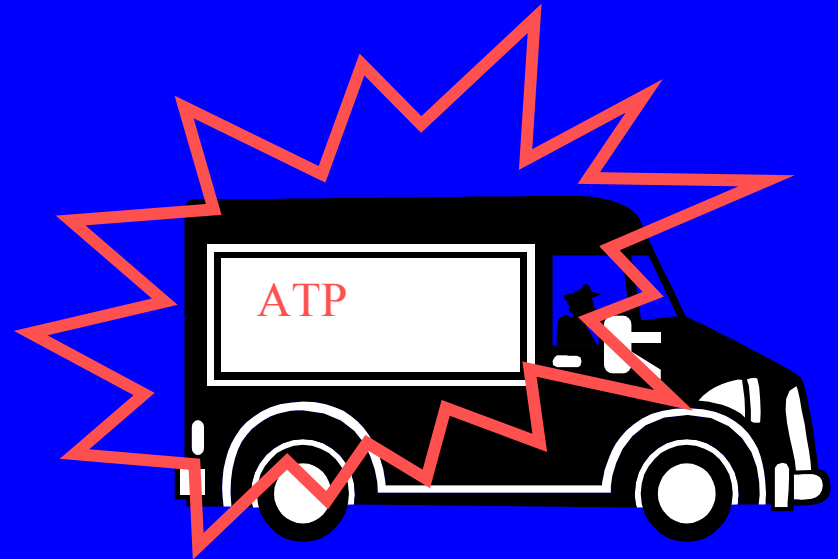
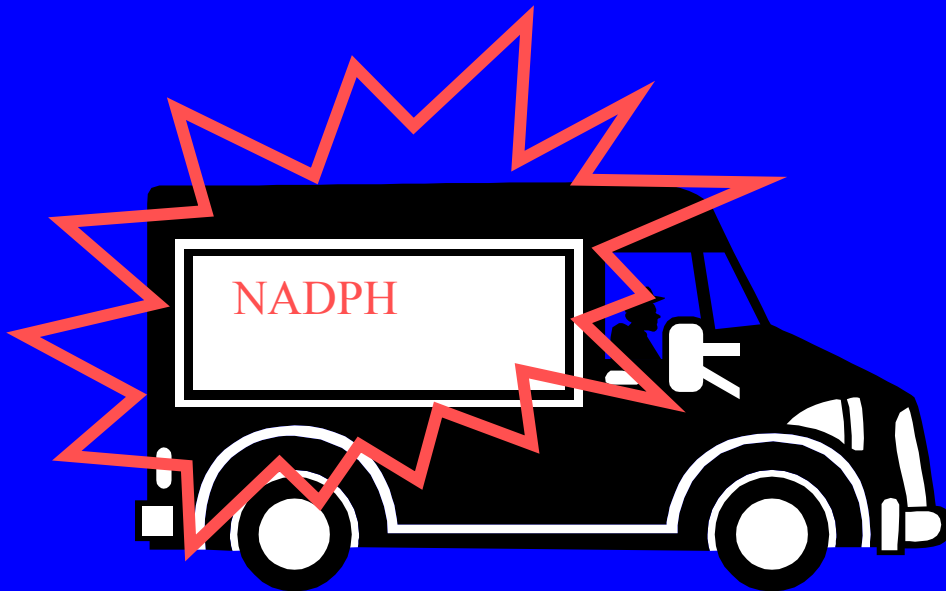
Production of the glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) molecule happens in two reactions.

The photosynthesis reaction can be seen as being as light dependent and light independent reactions.

energy is harnessed to make high

A. Light Dependent Reaction - light energy carrier molecules ATP and

NADPH.



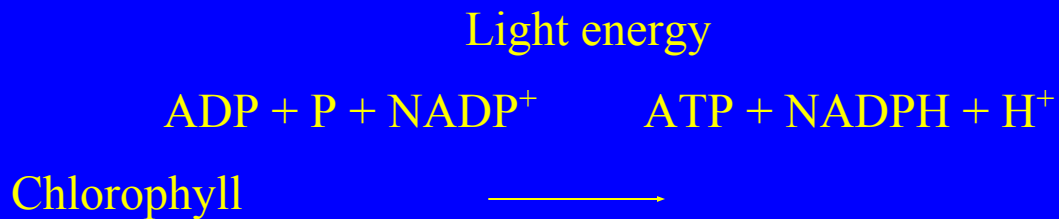
a. Photosystem II (Light Reaction)

Light energy is absorbed by chlorophyll and then

Part of the energy splits water into 2 Hydrogen (H^+) and an Oxygen (O_2) (photolysis). Oxygen is released from the plant and Hydrogen becomes attached to a carrier molecule called $NADP^+$ (nicotinamide adenine dinucleotide phosphate, a strong electron acceptor), which is reduced to form NADPH and carried to the dark reaction.

The other part of the energy is used to form a bond between ADP and phosphate to produce ATP.

ATP is needed in the dark (light independent) reaction for its available energy and NADPH is needed for its reducing power and hydrogen.

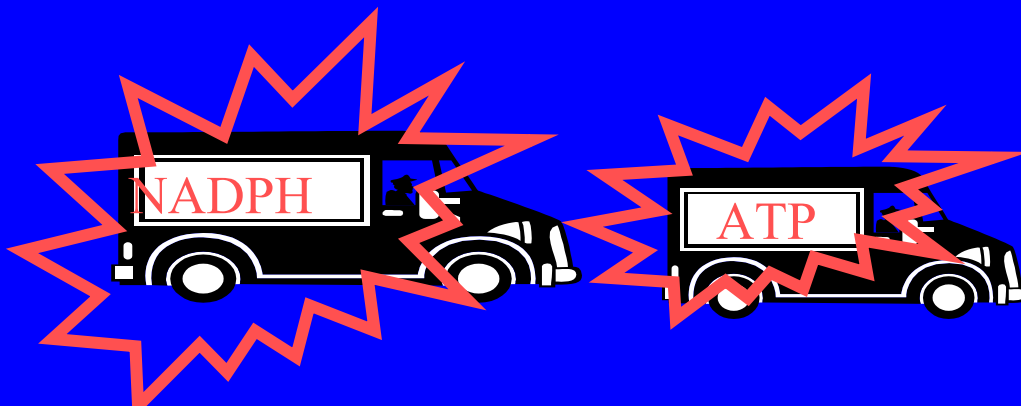


Photosystem 1 (Dark Reaction)

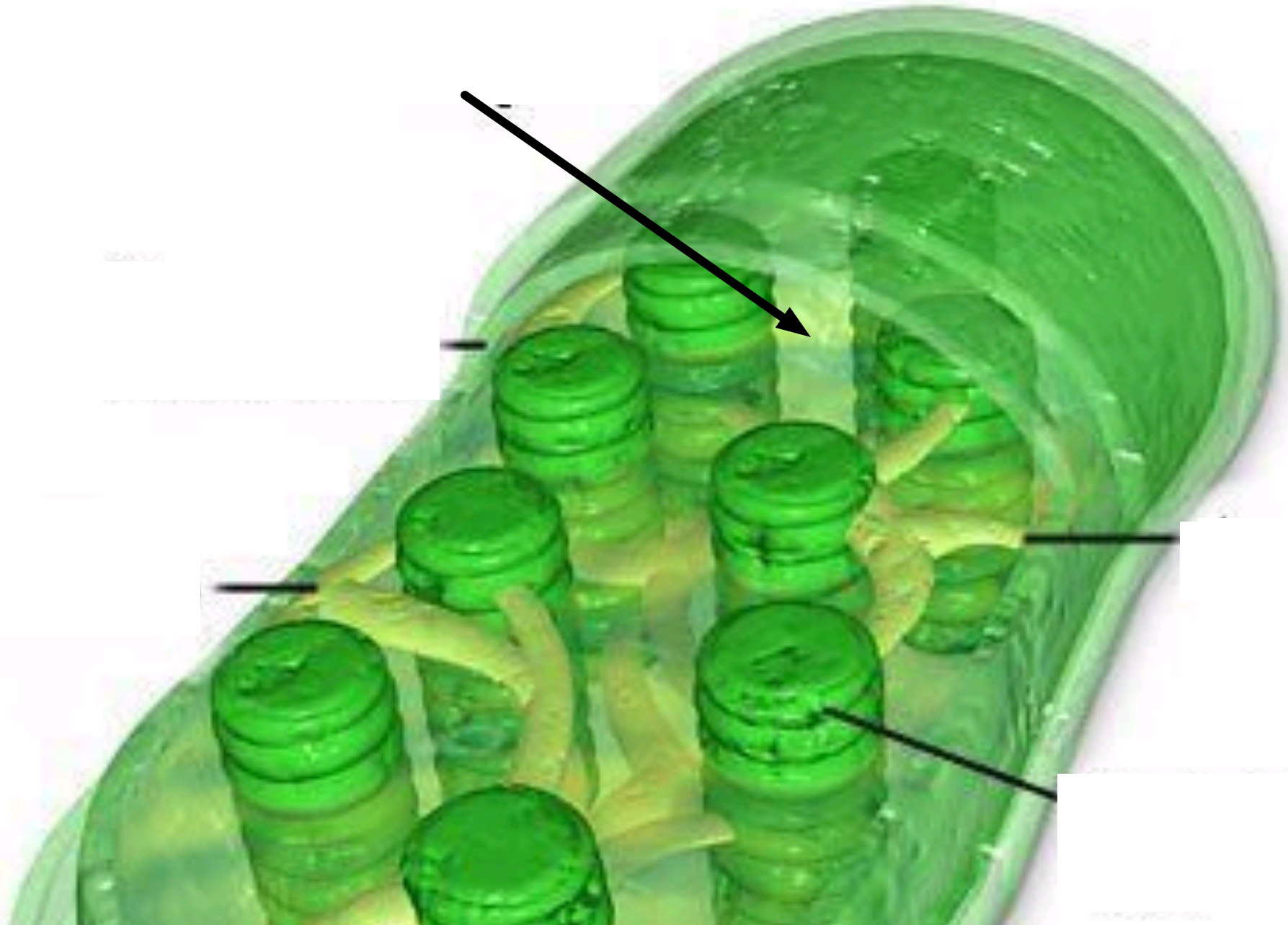
1. Energy rich ATP and NADPH
to build carbohydrate.
2. During the Calvin Cycle, CO₂
be “fixed” (made solid) to form

(aka The Calvin Benson Cycle)

are used as energy sources
from the atmosphere will
carb.



3. The reactions of the Calvin Cycle occur in the stroma.



Energy supplied by breakdown of ATP

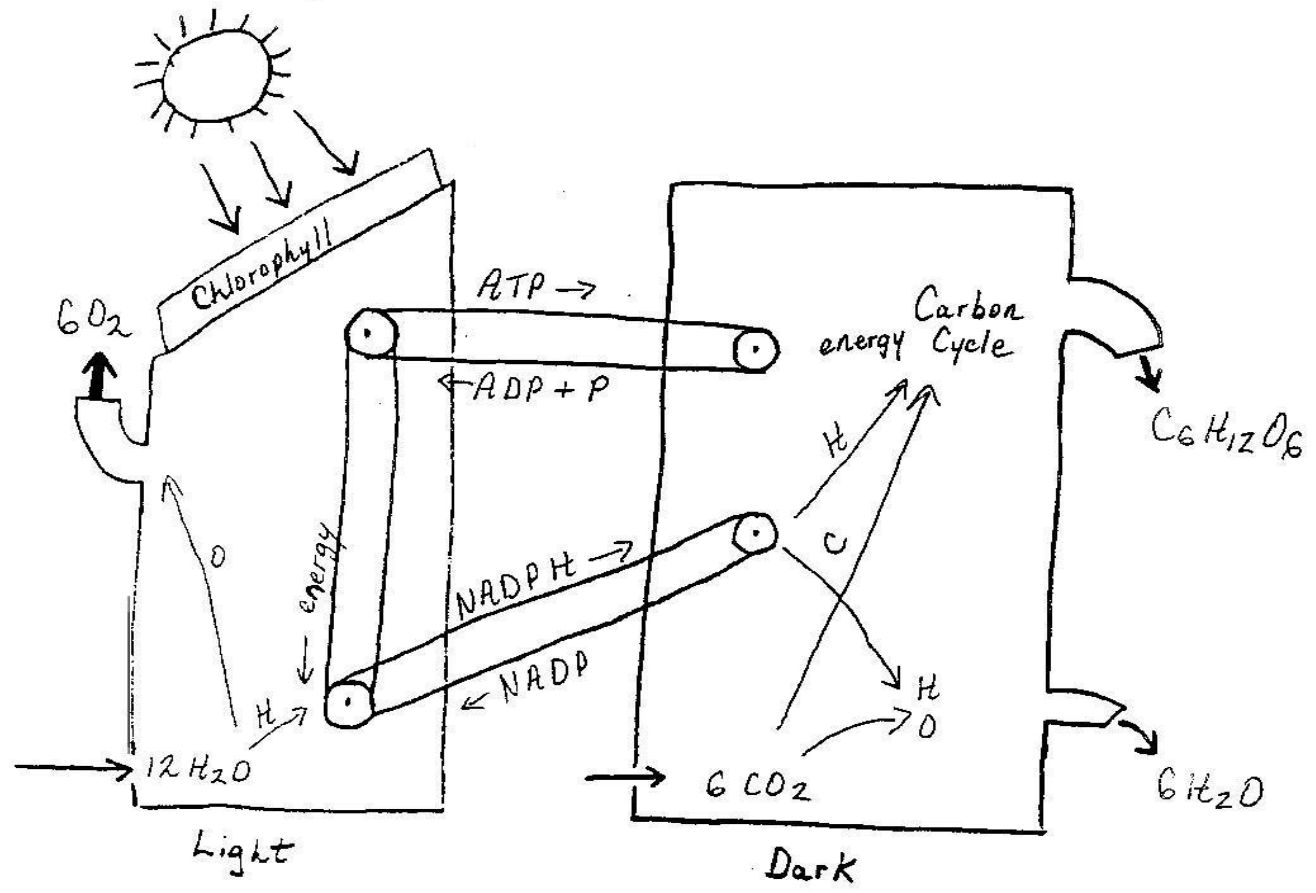
1. CO₂ taken up and rearranged in a succession of carbon compounds through the carbon cycle.
2. NADPH carrier molecules release hydrogen atoms.
 - a. 6 atoms of carbon, 12 atoms of hydrogen, and 6 atoms of oxygen combine to form glucose

b. the six remaining O atoms and
combine to form water
byproduct

12 remaining H atoms
which is released as a

Photosynthesis

Light/Dark Reactions





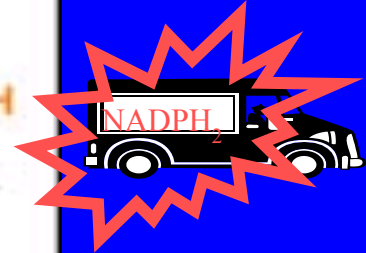
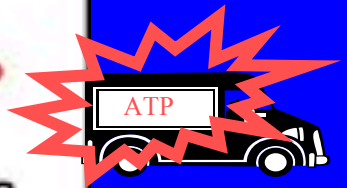
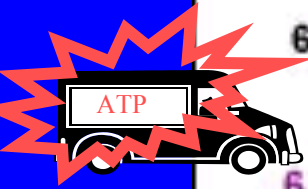
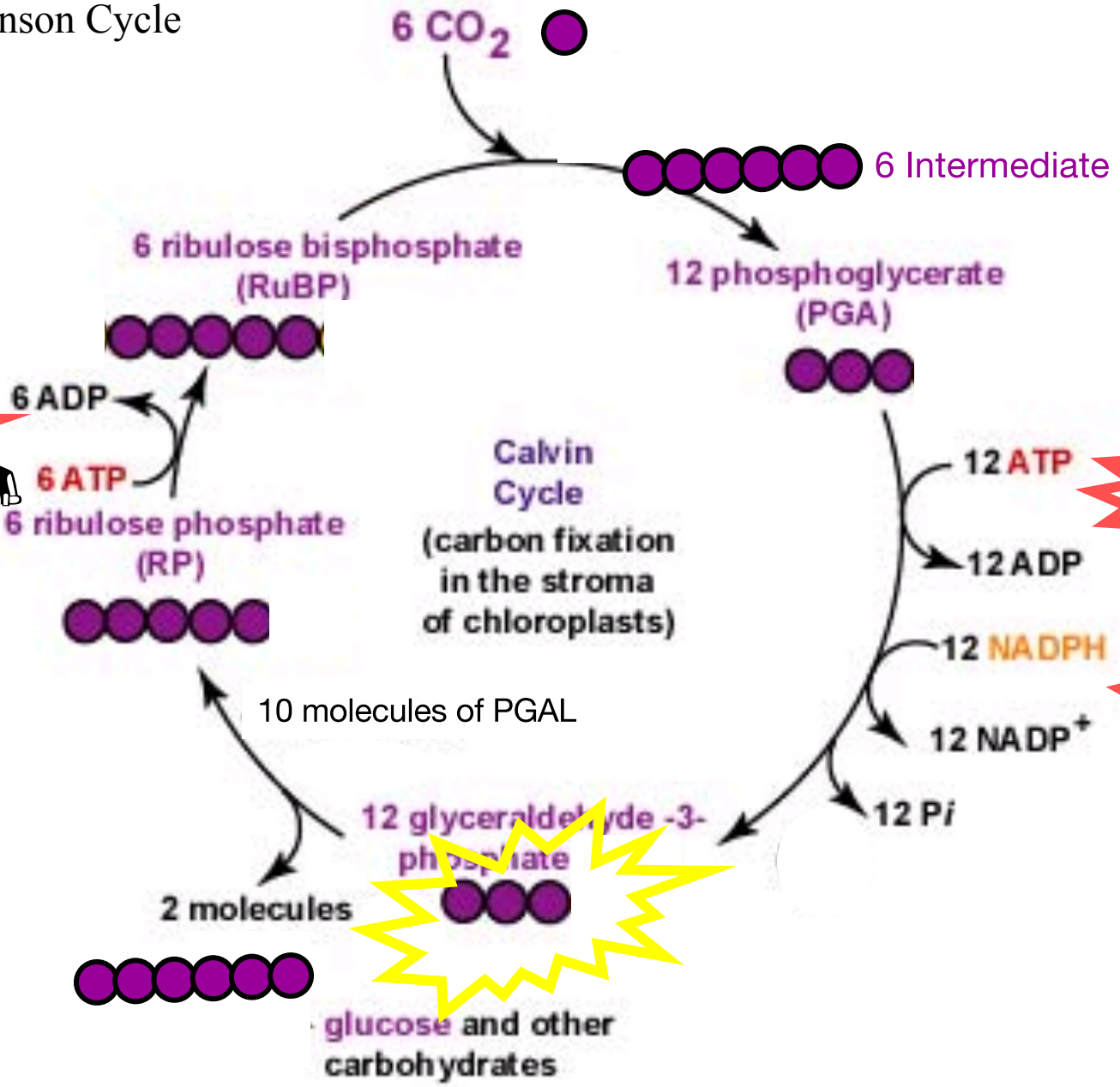
The Details

- a. 6 CO₂ from the air are combined with 6 molecules of RuBP (RuBP = Ribulose biphosphate = a five carbon sugar).
 - i. An enzyme called “Rubisco” makes this possible.

- b. The result is a 6 carbon intermediate molecule which is unstable.
- c. The 6 intermediates split into 12 molecules called PGA. Each is 3 carbons long.
Trivia: PGA = phosphoglycerate
- d. The energy from 12 ATP and 12 NADPH₂ are added to form 12 PGAL.
 - i. PGAL are energy rich
 - ii. PGAL = Think of L for “Light energy!”

- e. Two PGAL combine to form one glucose molecule.
- f. The 10 leftover PGAL are rearranged to form 6 RuBP using 6 ATP molecules as energy
- g. Repeat.

Calvin Benson Cycle



IV. The Metabolic Fates of Glucose

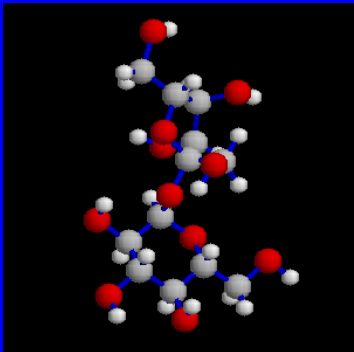
A. Once glucose is formed it may be:

1. Stored as pure glucose.

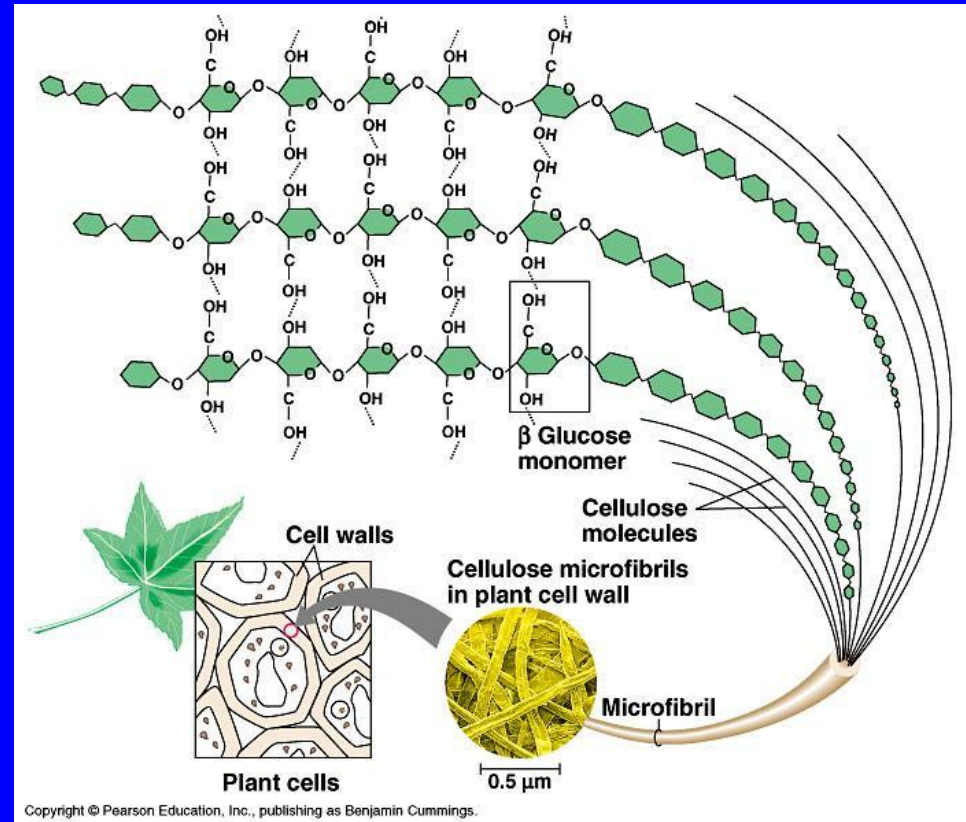
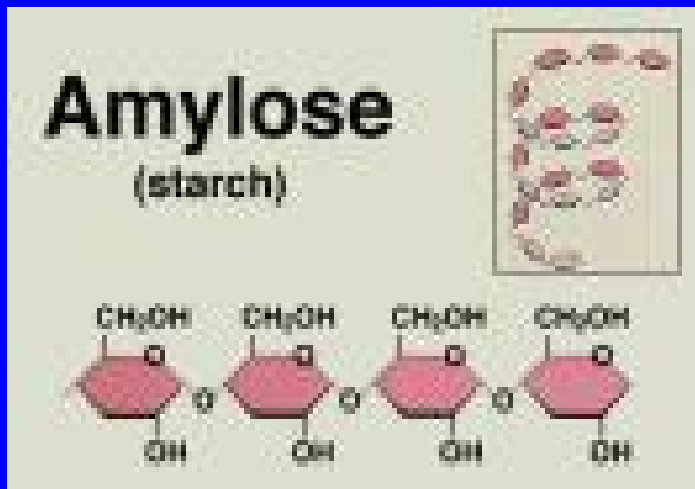


2. Combined with another sugar

to form a disaccharide, eg maltose



3. Combined with many sugars to form a polysaccharide.
 - a. Amylose (starch)
 - b. Cellulose (wood)



4. Oxidized (burned) to form ATP